SYNTHESIS OF 1-T₂-2-C¹⁴- AND OF 1-D₂-2-C¹⁴- TRANS-TRANS-FARNESYL PYROPHOSPHATE AND THEIR UTILIZATION IN SQUALENE SYNTHESIS

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Received February 8, 1961

We have presented evidence recently indicating that during biosynthesis of squalene from farnesyl pyrophosphate (FPP) by liver microsomes one hydrogen atom attached to C-1 of one of the two FPP molecules condensing to squalene is exchanged for a hydride ion derived from TPNH (Popják et al. 1961). This conclusion was drawn from the examination of the labeling of squalene biosynthesized from mevalonate-2-Cl4-5-D₂ and from the observation that during synthesis of squalene from biosynthetically prepared Cl4-FPP by liver microsomes in the presence of THO or T-labeled TPNH nearly one atom of labeled H was transferred to squalene only from the labeled TPNH and not from THO. We felt that further formal proof for the one exchanged H-atom referred to was required. We have therefore synthesized 1-T₂-2-Cl4- and 1-D₂-2-Cl4-trans-trans-farnesyl pyrophosphate from labeled trans-trans-farnesols and tested the preparations in biosynthetic experiments.

First, methyl farnesoate-2-C¹⁴ (3,7,11-trimethyldodeca-2,6,10-trienoate) was synthesized from trans-geranylacetone (1.94 g.) and methyl bromoacetate-2-C¹⁴ (1.53 g., sp. a. about 0.15 μ c./ μ mole) by the Reformatsky reaction, which gave methyl 3-hydroxy-3,7,11-trimethyldodeca-6,10-dienoate (1.86 g.). The ester of this hydroxy acid was dehydrated with POCl₃ (1.6 ml.) in pyridine (16 ml.) during 3 days at room temperature. After the usual extraction procedures 1.68 g. of methyl farnesoate-2-C¹⁴ was obtained which, according to gas-liquid radiochromatographic analysis (Popják et al. 1959), consisted of methyl cis-

trans-farnesoate (45%) and trans-trans-farnesoate (43%) and of some other unidentified radioactive impurities (12%). The <u>cis-trans</u> and <u>trans-trans</u> components were separated by preparative gas-liquid chromatography at 1970 using a 6 foot x l inch column packed with Celite 545 coated with 10% "Apiezon-L" vacuum grease (cf. Popják and Cornforth, 1960). The purity of the separated methyl <u>trans-trans-farnesoate-2-C¹⁴</u> was 95% by gas-liquid chromatography and was considered sufficient for further work.

 $\frac{\text{Trans-trans}}{\text{trans-trans-farmesol-l-T}_2-2-C^{14}}, \text{ and } \text{l-D}_2-2-C^{14} \text{ were prepared by reduction of methyl } \frac{\text{trans-trans}}{\text{trans-trans-farmesoate-2-C}^{14}} \text{ with T-labeled LiAlH}_4 \text{ and LiAlD}_4$ respectively. Experimental details are given for the T-labeled compound.

Methyl <u>trans-trans</u>-farnesoate-2- C^{14} (40.8 mg.) was diluted with unlabeled ester (367.6 mg.) (synthesized and purified as the C^{14} -compound) and added in ether (5 ml.) to a solution of tritio-LiAlH₄. (10 mg. LiAlT₄ + 85 mg. LiAlH₄) in ether (10 ml.) at -30° during 30 minutes, the reaction mixture being stirred at -30° for a further 90 minutes. After the usual working up and exchange of labile H with ethanol and H₂O, 369 mg. of farnesol-1-T₂-2- C^{14} were obtained, 95% of which consisted of the <u>trans-trans</u>-isomer (gas-liquid chromatography).

Trans-trans-farnesol-1-D2-2-C¹⁴ was made similarly from 506 mg. of methyl trans-trans-farnesoate-2-C¹⁴ and 422 mg. of LiAlD4 (99.2% D); yield 425 mg. of trans-trans-farnesol-1-D2-2-C¹⁴ (93% pure). The labeled farnesols were phosphorylated according to the procedure of Cramer and Böhm (1959); 333 mg. of the T-farnesol yielded 100 mg. of the monophosphate (cyclohexyl-ammonium salt) and 301 mg. of the pyrophosphate (Li-salt); 403 mg. of the D-farnesol gave 87 mg. of the monophosphate and 127 mg. of the pyrophosphate together with some further unidentified products. The synthetic FPP preparations (pyrophosphate-P: 14.0%, theoretical for Li3-salt 15.5%; total-P 14.9%) behaved as the biosynthetic specimens in that they cleaved rapidly

^{*} Specific activity unknown as the suppliers informed us after the experiment was completed that they had reasons to believe that the preparation was unsatisfactory. The radioactive yield of the T-labeled farnesol was approximately 1/100th of that anticipated.

at room temperature in acid solution (pH < 2) releasing inorganic pyrophosphate with concommittant allylic rearrangement of most of the farnesol to nerolidol (Lynen et al. 1958; Goodman and Popják 1960). Hydrolysis with intestinal alkaline phosphatase at pH 8.7, on the other hand, gave trans-trans-farnesol (85%), cis-trans-farnesol (5.3%), nerolidol (0.55%) and four further small components not identified.

Both the $1-D_2-2-C^{14}$ and the $1-T_2-2-C^{14}$ -FPP preparations were tested as substrates in the anaerobic squalene synthetase system of rat liver microsomes in standard 1 ml. incubations described in detail previously (Goodman and Popjak, 1960, Popjak et al. 1961) and were converted to squalene. The $1-T_2-2-C^{14}$ -farnesol contained 89,100 dopomo of T and 30,050 dopomo of C^{14} per umole, and the squalene biosynthesized from the 1-T2-2-C14-FPP 2330 d.p.m. of T and 1018 d.p.m. of C^{14} . Calculating the yield of squalene from the C^{14} counts, there were synthesized $1018/(30,050 \times 2) = 0.0169$ µmoles of squalene, within the range of the biosynthetic activity usually observed in 1 ml. incubations (cf. Popják et al. 1961). It is obvious that the T/C 14 ratio in the FPP and squalene biosynthesized from it were different. Taking this ratio in the FPP to have been 1.00, the ratio in the squalene was < 0.77 indicating the loss of about one labeled H-atom from one of the two FPP molecules (the theoretical ratio for the loss of one labeled H-atom out of four is 0.75) in complete agreement with our previous results (Popjak et al. 1961). It is highly probable that the loss of the one hydrogen atom occurs from an intermediate formed after the condensation of two sesquiterpenoids (cf. Popjak et al. 1961) and that the removal of the H-atom is stereospecifically determined and hence no "isotope effect" should be discernible as the two H-atoms of C-l of FPP must have been labeled to the same extent.

The data from the experiment with $1-D_2-2-C^{14}$ -FPP are relevant at present only to the extent that the chemically synthesized specimen (like the $1-T_2-2-C^{14}$ -FPP) acted as substrate for the biosynthesis of squalene (in a 1 ml. incubation 0.04 µmole squalene being synthesized) providing a formal proof for

the identity of the immediate precursor of squalene deduced previously from the study of trace (or μ mole) amounts of biosynthetic material (Lynen et al., 1958; Goodman and Popják, 1960).

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